

Structure of 3-Amino-5-hydroxyquinoline-7-sulfonic Acid Monohydrate

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(Received 19 March 1986; accepted 25 June 1986)

Abstract. $C_9H_8N_2O_4S.H_2O$, $M_r = 258.25$, monoclinic, $P2_1/c$, $a = 9.912$ (1), $b = 14.775$ (1), $c = 7.093$ (1) Å, $\beta = 98.585$ (5)°, $V = 1027.1$ (2) Å³, $Z = 4$, $D_x = 1.670$ (1) g cm⁻³, Mo $K\alpha$, $\lambda_{\alpha_1} = 0.70930$ Å, $\mu = 3.1$ cm⁻¹, $F(000) = 536$, $T = 295$ K, $R = 0.033$ for 1188 reflections. The molecule exists in the crystal as a zwitterion, with an approximately planar quinoline ring system and normal dimensions; molecules are linked by hydrogen bonds *via* the water molecule.

Introduction. Cleavage of western hemlock bark tannins by treatment with ammonium sulfite in concentrated ammonium hydroxide solution (Laks, 1984) yields products which may be useful as adhesives in the forest products industry. Chemical methods did not define the molecular structure of one of these products unambiguously, and the present X-ray study was designed to establish the structure.

Experimental. Golden plates elongated along *c*, dimensions 0.06 × 0.17 × 0.6 mm, faces {100}, {010}, {011}, Enraf–Nonius CAD-4F diffractometer, lattice parameters from 25 reflections with $\theta = 17.4$ – 21.4 °. Intensities for $\theta \leq 27.5$ °, hkl : –12 to 12, 0 to 19, 0 to 9, ω – 2θ scan, ω -scan width $(0.65 + 0.35 \tan \theta)$ ° at 0.9–10° min⁻¹, extended 25% on each side for background measurement, three standard reflections (no decay), L_p and absorption corrections, transmission factors 0.945–0.980. 2335 reflections measured, 1188 with $I \geq 3\sigma(I)$, where $\sigma^2(I) = S + 4(B_1 + B_2) + (0.04S)^2$, $S = \text{scan}$, B_1 and $B_2 = \text{background counts}$. Structure by Patterson and Fourier methods, refined by full-matrix least squares on F , H atoms from a difference map, refined with isotropic thermal parameters, $w = 1/\sigma^2(F)$, scattering factors from *International Tables for X-ray Crystallography* (1974), locally written programs, and locally modified versions of *ORFLS* and *ORFFE* (Busing, Martin & Levy, 1962, 1964), and *FORDAP* (A. Zalkin, unpublished). Final $R = 0.033$, $wR = 0.041$ for 1188 reflections, $S = 1.61$, 194 parameters, $R = 0.110$ for all 2335 reflections,

Table 1. Final positional (fractional × 10⁵, H × 10⁴) and isotropic thermal parameters ($U \times 10^3$ Å²) with *e.s.d.*'s in parentheses

	$U_{eq} = \frac{1}{3}$ trace of diagonalized U tensor.			U_{eq}/U_{iso}
	<i>x</i>	<i>y</i>	<i>z</i>	
S	19693 (7)	40753 (5)	37637 (12)	32
O(1)	13256 (21)	36781 (16)	52731 (33)	45
O(2)	11333 (22)	39777 (18)	19171 (32)	53
O(3)	24300 (22)	49930 (15)	41811 (35)	50
O(4)	44589 (24)	10249 (14)	36553 (34)	41
O(5)	19889 (26)	3150 (16)	36393 (45)	45
N(1)	69760 (25)	36553 (19)	27965 (39)	35
N(2)	91655 (30)	17460 (27)	23106 (47)	51
C(2)	80595 (31)	31757 (23)	25288 (45)	37
C(3)	80307 (28)	22357 (22)	25825 (40)	34
C(4)	68418 (29)	18231 (23)	29421 (44)	32
C(4a)	56980 (26)	23389 (19)	32100 (37)	27
C(5)	44487 (28)	19346 (19)	35530 (40)	30
C(6)	33596 (27)	24746 (20)	37606 (42)	30
C(7)	34605 (27)	34225 (20)	36331 (39)	28
C(8)	46397 (28)	38410 (22)	33472 (43)	31
C(8a)	57616 (27)	32890 (20)	31303 (39)	28
H(O4)	3733 (37)	836 (25)	3724 (51)	51 (12)
H(O5a)	1613 (49)	535 (38)	4785 (71)	115 (19)
H(O5b)	1577 (41)	620 (33)	2712 (57)	75 (16)
H(N1)	7053 (36)	4220 (28)	2894 (53)	60 (12)
H(N2a)	9865 (38)	2042 (27)	1770 (56)	70 (13)
H(N2b)	9042 (46)	1140 (37)	2401 (65)	97 (18)
H(2)	8889 (30)	3505 (21)	2309 (39)	37 (9)
H(4)	6748 (28)	1238 (24)	2834 (44)	37 (9)
H(6)	2574 (33)	2206 (23)	3882 (45)	43 (9)
H(8)	4671 (25)	4433 (21)	3248 (40)	23 (7)

$\Delta/\sigma = 0.009$ (mean), 0.095 (max.), max. final difference density –0.25 to +0.36 e Å⁻³. Thermal parameters interpretable in terms of rigid-body motion, bond lengths corrected for libration.

Discussion. Atomic positional parameters are given in Table 1.† The molecule (Fig. 1) contains a quinoline ring system, with amino, hydroxy, and sulfonate substituents, and exists in the crystal in a zwitterionic form, with the acidic H atom bonded to the ring N

† Lists of anisotropic thermal parameters, bond lengths and angles involving H atoms, torsion angles, and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 43187 (16 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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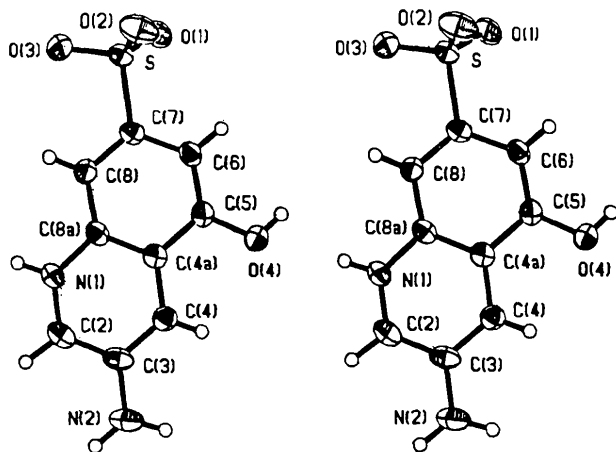


Fig. 1. Stereoview of the 3-amino-5-hydroxyquinoline-7-sulfonic acid molecule.

atom. The quinoline ring system is slightly folded, with an angle of 1.5° between the almost planar six-membered rings; the NH_2 group is slightly non-planar [N(2) is displaced by $-0.083(3)$ Å from the plane of the three bonded atoms, with this plane rotated 8.8° out of the attached ring plane], and S is $0.118(1)$ Å from the plane of the six-membered ring to which it is bonded. Bond lengths and angles (Table 2) are generally close to those in related molecules, e.g. 8-hydroxy-2-methylquinoline-5-sulfonic acid monohydrate (Merritt & Duffin, 1970). The molecules are linked by a system of hydrogen bonds (Table 2) involving all the active H atoms.

We thank Drs R. W. Kennedy and J. P. Kutney for suggesting the problem, the Natural Sciences and Engineering Research Council of Canada for financial support, and the University of British Columbia Computing Centre for assistance.

Acta Cryst. (1986). C42, 1800–1802

Structure of 8-Bromo-6-endo-methylisofenchone*

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(Received 19 March 1986; accepted 25 June 1986)

Abstract. $\text{C}_{11}\text{H}_{17}\text{BrO}$, $M_r = 245.16$, orthorhombic, $P2_12_12$, $a = 10.957(1)$, $b = 22.409(2)$, $c =$

* (1*S*,4*S*,5*S*,6*S*)-5-Bromomethyl-1,5,6-trimethylbicyclo[2.2.1]-heptan-2-one.

Table 2. Bond lengths (Å) and angles ($^\circ$), with e.s.d.'s in parentheses

	Uncorr.	Corr.		Uncorr.	Corr.
S—O(1)	1.450 (2)	1.461	C(3)—C(4)	1.384 (4)	1.387
S—O(2)	1.449 (2)	1.460	C(4)—C(4a)	1.402 (4)	1.405
S—O(3)	1.447 (2)	1.459	C(4a)—C(5)	1.428 (4)	1.431
S—C(7)	1.779 (3)	1.787	C(4a)—C(8a)	1.407 (4)	1.412
O(4)—C(5)	1.346 (4)	1.349	C(5)—C(6)	1.368 (4)	1.370
N(1)—C(2)	1.324 (4)	1.326	C(6)—C(7)	1.408 (4)	1.413
N(1)—C(8a)	1.372 (4)	1.375	C(7)—C(8)	1.364 (4)	1.367
N(2)—C(3)	1.375 (4)	1.377	C(8)—C(8a)	1.406 (4)	1.408
C(2)—C(3)	1.390 (5)	1.395			
O(1)—S—O(2)	111.84 (14)		C(4)—C(4a)—C(8a)	119.7 (3)	
O(1)—S—O(3)	112.86 (14)		C(5)—C(4a)—C(8a)	118.0 (3)	
O(1)—S—C(7)	106.24 (14)		O(4)—C(5)—C(4a)	115.3 (3)	
O(2)—S—O(3)	113.49 (15)		O(4)—C(5)—C(6)	125.2 (3)	
O(2)—S—C(7)	105.45 (14)		C(4a)—C(5)—C(6)	119.5 (3)	
O(3)—S—C(7)	106.23 (13)		C(5)—C(6)—C(7)	120.6 (3)	
C(2)—N(1)—C(8a)	124.4 (3)		S—C(7)—C(6)	117.9 (2)	
N(1)—C(2)—C(3)	120.7 (3)		S—C(7)—C(8)	119.9 (2)	
N(2)—C(3)—C(2)	120.1 (3)		C(6)—C(7)—C(8)	122.1 (3)	
N(2)—C(3)—C(4)	122.1 (3)		C(7)—C(8)—C(8a)	117.6 (3)	
C(2)—C(3)—C(4)	117.8 (3)		N(1)—C(8a)—C(4a)	116.5 (3)	
C(3)—C(4)—C(4a)	120.9 (3)		N(1)—C(8a)—C(8)	121.2 (3)	
C(4)—C(4a)—C(5)	122.3 (3)		C(4a)—C(8a)—C(8)	122.2 (3)	

Hydrogen bonds

	X...O	X—H		
O(5)—H ₁ ...O(2)	2.794 (3)	1.00 (5)	1.80 (5)	172 (5)
O(5)—H ₂ ...O(1)	2.807 (4)	0.85 (4)	2.00 (4)	158 (4)
O(4)—H...O(5)	2.662 (3)	0.78 (4)	1.89 (4)	174 (4)
N(1)—H...O(5)	2.901 (4)	0.84 (4)	2.24 (4)	136 (3)
N(1)—H...O(3)	2.925 (4)	0.84 (4)	2.37 (4)	124 (3)
N(2)—H ₁ ...O(1)	2.827 (4)	0.95 (4)	2.20 (4)	123 (3)
N(2)—H ₂ ...O(3)	3.135 (5)	0.91 (5)	2.40 (5)	138 (4)

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